

PHARMACEUTICAL DOSAGE FORMS

Tablets

SECOND EDITION, REVISED AND EXPANDED

In Three Volumes

VOLUME 2

EDITED BY

Herbert A. Lieberman

H. H. Lieberman Associates, Inc.
Consultant Services
Livingston, New Jersey

Leon Lachman

Lachman Consultant Services
Westbury, New York

Joseph B. Schwartz

Philadelphia College of Pharmacy and Science
Philadelphia, Pennsylvania

MARCEL DEKKER, INC.

New York and Basel

Library of Congress Cataloging-in-Publication Data
(Revised for vol. 2)

Pharmaceutical dosage forms--tablets.

"In three volumes."

Includes bibliographical references.

1. Tablets (Medicine) 2. Drugs--Dosage forms.

I. Lieberman, Herbert A.

II. Lachman, Leon

III. Schwartz, Joseph B.

[DNLM: 1. Dosage forms. 2. Drugs--administration &
dosage. QV 785 P535]

RS201.T2P46 1989

615'.191

89-1629

ISBN 0-8247-8044-2 (v. 1 : alk. paper)

This book is printed on acid-free paper.

Copyright © 1990 by MARCEL DEKKER, INC. All Rights Reserved

Neither this book nor any part may be reproduced or transmitted in any form or by any means, electronic or mechanical, including photocopying, microfilming, and recording, or by any information storage and retrieval system, without permission in writing from the publisher.

MARCEL DEKKER, INC.

270 Madison Avenue, New York, New York 10016

Current printing (last digit):

10 9 8 7 6 5 4 3 2 1

PRINTED IN THE UNITED STATES OF AMERICA

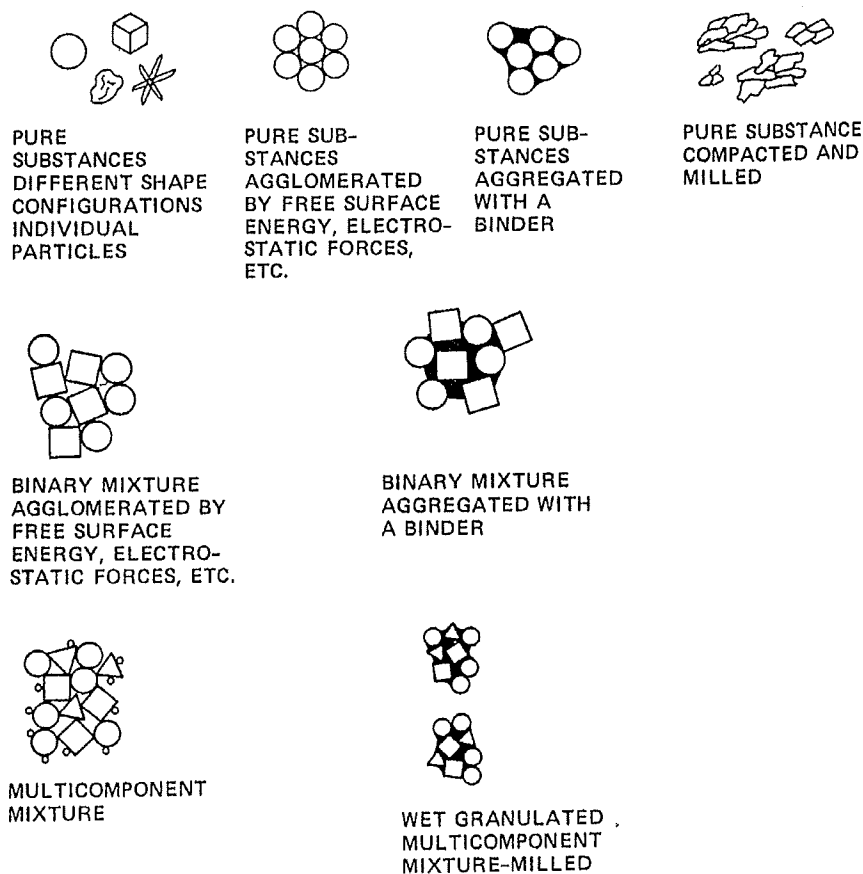


Figure 21 Several different types of particles encountered in tablet granulation dry blending.

may be found in the reference text: *Handbook of Pharmaceutical Excipients* [33].

Large (sieve size range >60 mesh) dry particles have a tendency to flow better than the smaller dry particles, because they have greater mass. Smaller particles (<100 mesh) may create mixing problems because surface areas are very great, and may give rise to strong electrostatic forces as a result of processing and/or inter-particle friction from movement. These forces may prevent the desired distribution of these smaller particles throughout a mixture because of fine particle agglomeration.

As the particle size approaches 10 μm and below, weak polarizing electrical forces called van der Waals forces or cohesive forces also begin to affect the flow of the powder. Both van der Waals and electrostatic forces usually inhibit powder flow through particle agglomeration as mentioned above. However, in some instances improved flow results because

Table 5 Effect of Particle Size on Powder Flow

| Particle size | Type of flow ^a | Reason |
|--|---|---|
| 200-250 μm (10-60 ^b mesh) | Flow is usually good if shape is not interfering | Mass of individual particles is relatively large |
| 250-75 μm (60 mesh-200 μm) | Flow properties may be a problem with many pure substances and mixtures | Mass of individual particles is small and increased surface area amplifies effects of surface forces |
| <100-75 μm | Flow becomes a problem with most substances | Cohesive forces or free surface energy forces are large as well as static electrical forces relative to particle size |

^a Assume particle shape is constant and does not interfere with flow.

^b U.S. standard mesh size.

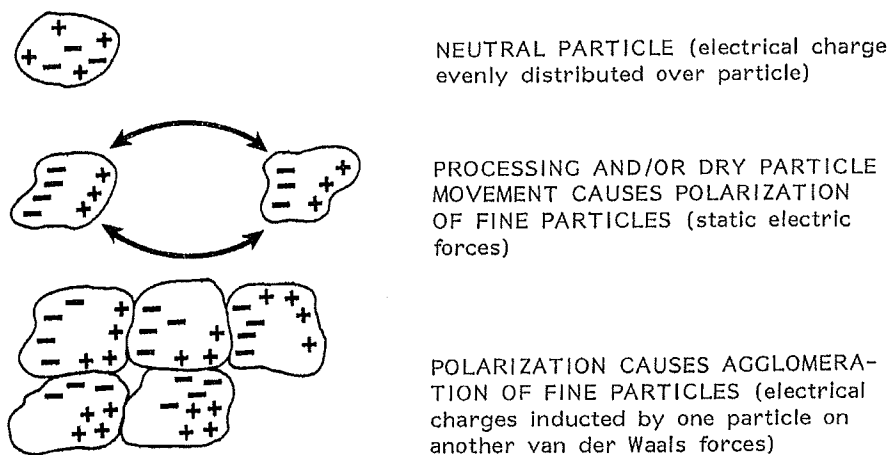


Figure 22 Effect of electrical forces on fine particles.